NOTES ON GEO N-1 GEOPHYSICAL WELL LOGS

GL014416

CALIPER LOG

The caliper log shows a highly uniform hole diameter and shape over most of the length of the hole. There are a few areas of enlargement that correspond for the most part to flow boundaries or to suspected fractures. The interval immediately adjacent to the bottom of the rain curtain at 3260 is enlarged to 5 in, and the largest washout occurs at 3429, where the hole diameter is 9.7 in.

TEMPERATURE LOGS

There were two temperature logs run in the well. The first was run on 2 Nov 85 during the course of the Dresser Atlas logging. The second temperature log was run on 9 Nov 85 by Geotech Data of Poway, Ca. This second log used a tool having an apparent sensitivity of about 0.01 C, which is believed to be considerably better than the sensitivity of the usual logging tool temperature probe. In addition, the second tool is likely to be better calibrated. The last fluid circulation is estimated to have occurred when the black iron pipe was set, which was completed on 8 Nov 85, one day earlier. Neither of the temperature logs is an equilibrium log. Nevertheless, they do yield data of significance.

The upper 200 ft of the hole shows temperatures about 45 F, which decrease downward to about 41 F and then begin a very show buildup such that at 2000 ft temperature is 42 F, and at 3000 ft temperature is 51 F. Between 3260 and 3300 ft, there is a very rapid increase in temperature to 103 F, and thereafter a more uniform gradient to a temperature of 160 F at 4000 ft. This area is in a fairly massive dacite unit. The abrupt temperature increase apparently signifies the bottom of the level of cold water circulation called the rain curtain. The average gradient ft corresponds to 115 C/km. below 3300 Average thermal conductivities for the rocks below 3300 ft are 4.3 mcal/cm-secdeg C, so that the indicated heat flow is about 5 HFU. We must remeber that the temperature profile is not equilibrium, however, and this heat flow value is only an indication of the true value below 4000 ft.

GAMMA RAY LOG

This log shows counts in standard API units. Basalt and andesite flows are fairly uniform, but some individual flows can be differentiated. Typical values of the basalts are 20-40 API units. A dacite ash at 1982 is clearly delineated. Several thin clay-altered units below about 3100 ft show high response. Dacite flows at 3211-3330 ft and 3708-4000 ft show increased response, averaging 130 API units. This log appears to be successful at differentiating the felsic and/or altered units from the basalt and basaltic-andesite flows.

ELECTRIC LOGS

The electric logs comprise an SP log, a 16-in short normal resistivity log and an induction log. The SP log was off scale for much of the upper part of the hole and appears to be of limited use for quantative interpretation in any case. It will not be discussed further. The resistivity of the borehole fluid at the time of the logging was not measured, and so the interpretations that can be placed on the resistivity and induction logs are somewhat compromised. The induction log is useful of the two remaining logs the more in yielding representative values of resistivity for the formations because of its greater depth penetration.

and the second second second

er i Stre

Both the resistivity and induction logs indicate the presence of conductive horizons below a depth of about 2800 ft. The conductors become more numerous and of higher conductivity down hole. The average resistivity of the upper 2000 ft of the hole is 50-70 ohm-m. Below this there is a systematic decrease in resistivity with depth. Below 2800 ft, there are at lease 15 separately identifiable horizons having resistivities below 10 ohm-m, and one horizon has a resistivity value of 1.2 ohm-m. The thickness of these horizons varies from a few feet to a few tens These conductive horizons correspond, for the most of feet. part, to clay-altered basaltic ash and felsic tuff units. The chief alteration type is smectite which has apparently developed at low temperature.

The conductive horizons observed on the electric logs are to be responsible for the occurrence of anomalies in believed interpreted conductivity on surface TDEM surveys reported by Dave Fitterman of the USGS. The surface surveys indicate a widespread area underlain by conductive horizons around the Newberry volcano. Part of the anomaly must correspond to the hightemperature hydrothermal system found in USGS Newberry-2 , which was drilled in the caldera. However, part of the anomaly must also correspond to the conductors found in GEO N-1.

ACOUSTIC LOGS

Two acoustic logs were run in the well--an acoustic velocity log and an acoustic fraclog. Both of these logs are useful in detecting flow boundaries and differentiating areas of uniform, probably low porosity/permeability from porous/permeable horizons that correspond for the most part to flow boundaries and, to a lesser extent, to fractures.

INTERPRETATION OF THE LOGS

These logs will be very useful when correlated with the core in calibrating log response in this sequence of basalt and basaltic-andesite flows with separate ash and tuff units. UURI is involved with this work at the present time. We plan to make detailed log correlations with the core, make such measurements as resistivity and perhaps IP effect on selected core specimens, and study cross plots.

DEPARTMENT OF ENERGY CASCADES DEEP GEOTHERMAL GRADIENT DRILLING PROGRAM

OBJECTIVE:

Sponsor research to characterize the deep hydrothermal regime of the Cascades in order to define its geothermal potential.

PROGRAM MANAGEMENT:

DOE/Idaho Operations Office

TECHNICAL COORDINATION:

University of Utah Research Institute

MECHANISM:

and the states of the second

Solicitations for Cost-Shared Drilling

STATUS

¢	GEO Operator	Corp.				
	N-1	(4000')	Completed 10/20/85 Data and samples open filed by UURI Feb 86			
	N-3	(4000')	Spud 6/1/86			
☆ Thermal Power Company						
	Clackamas	(5000')	Spud 6/1/86			
✡	Blue Lake Geot	thermal				
	Blue Lake	(4000')	Spud ?			
✡	New Solicitation: May 30, 1986					
	Two additional holes anticipated					

1





Priest et al., 1983

100 mW/m² 80 mW/m²



DOE GEOTHERMAL CASCADES DRILLING PROGRAM



DATA ON GEO N-1

and the state of

8/24/85 10/20/85 Tonto >90% \$96/ft. \$66/ft. 69ft./day 4550 ± ft. 0-4000 ft. 1 1/2" iron pipe to T.D.

Geophysical Well Logs GEO N-1

(Dresser Atlas, logged 11/2-3/85 unless noted)

Temperature ______ Caliper Gamma Ray Spontaneous Potential Resistivity Induction Acoustic Acoustic Fraclog DA

Geotech Data 11/9/86 4-arm

large portions off scale 16 in. short normal

Borehole compensated Borehole compensated





2

COREHOLE GEO N-1





DOE SUPPORTED SCIENTIFIC STUDIES

University of Utah Research Institute

Oregon Department of Geology and Mineral Industries

Southern Methodist University

U.S. Geological Survey

GEO Operator Corp

Thermal Power Company

Blue Lake Geothermal

and the second second

1 834

OPPORTUNITIES

- Holes available for experiments for one year following completion
- Splits of core stored at UURI available for study

* Geophysical well logs







	Sounding	hyper	Lower	111 -	
	ا م مراجع می اور این	^		_ wable Layer 1	aur L
· · · · · · · · · · · · · · · · · · ·		Michers	·	Anohun p	
·	19	130	1652	465 509	43
- 1 -	20		84.8	63 9 375	26
	21		4992	504 300	34-
	22	165	1689	5 3 3 6 9	28
 	2-3	813	953		53
	24	741	483		4 6
	25	120	2770	581 280	30
	26	150	2232	626 388	33
	27	122	1850	575 218	ZG
• •• •• •• ••	28	119	1.293	645 268	21
	29	791	413		23
	30	863	370		27
	31	100	1015	520 403	57
: 	32	793	456		40
	33	648	1364		3-ک
	34	90	775	573 2-91	39
	35	654	1424		67
4	: 36	692	2248		_د
••••••••••••••••••••••••••••••••••••••	37	726	2722		20
	38	100	E0 9	550 216	
ر: ۲۰ نوب	39	130	1570	551 289	25
	FO	612	1956		7 :
: ; ; ;	41	582	1368	······	20

2800 fir 4000' Aurage P

Internal p 3800 - 20 10× 8 3300-20 40 2800-20 47 1035 -40 25 - 40 10 -40 10520 -60 15 -60 8 ~60 .30 -80 40 -80 5 -80 15 -LOU LOKED -100 18 -100 20 3400-20 12435 8×15 3900-20 30 2900 - 20 35 -40 30 -90 95 -40 3 -60 35--60 25 -60 . 37 -80 25 -80.35 -80 18 -100 10×50 -100 30 -100 42 4000 -20 35 3520-20 50 -90 27 -40 25 -60 95 -60 30 -80 40 -90 25 -100 90 -100 15 5 pchi = 28 A-m 20 . 30 3600 - 20 30 $\exists \iota$ - 90 10-6 - 40.35 5 he -60.20 -60 20 -80 10×12 05×00 08--80 18 -100 25 -100 30 3700-20 5-- 20 36 320 - 70 40 - 40 30 -60 38 -60 45

-100 42 -100 40 -10030

	East Lake Hot Springs	East Lake Hot Springs	Paulina Hot Springs	Little Crater Campground Warm-Well
Date of sampling	1973	8/1975	7/1977	8/1975
Specific conductivity	396	767	- /	900 M. M.
Temp°C	62.0	49.	- /	35.5
трН	6.49	6.42	6.82 /	6.46
/ SiO ₂	36.	100.	205.	161.
Na +	32.	53.	140.	83.+
KT	3.8	_	17.	(/.10)
Ca + +	38.	70.	56.	¥ 54.++/
Mg+f	16.	34.	60.	1 48.++ /
HCO3 -	184	547	856	679.
S04 =)-	58.	28.	~1.	14-61-5
	0.4	1.7	6.0	5:1
F_	0.2	0.16	0.57	0.6
В	0.93	1.1	0.87	2.5
Li	0.01	0.04	0.22	0.12
Rb	<0.02	0.03	0.04	0.02
Cs	<0.1	<0.1	<0.1	<0.1
Sr	0.14	-	-	-
A1	-	0.008m	-	0.002m 🗼
Fe	<0.02	0.66	-	4.
Mn	0.10	0.90	-	0.25
Hg	0.0003	-	-	0.0001
* Concentrations in mg/l;	, m denotes mono	meric aluminum.	Reg 11-12-10	csturty = @96°F
$\frac{200 \mu who}{cm} = \frac{200 \mu S}{cm}$ $= \frac{200 \times 10^{-6} \times 10^{2} S}{M}$	<u>16 л</u> -	m @ 200 ppu,	50°F	「「」、「」、「」、「」、」、
= 200 x10 mg	-m @ 96°F 40	- at 50°F,	R= 221-4	

Table 5.1. Chemical composition of thermal waters at Newberry volcano* (Source: Mariner and others, 1980)

TDS

wrater Desistanty Colculations 1. For Little Cratir es warm ail _____ (a) = 900 <u>pes</u> measured $= 900 \times 10^{-4} \text{ S} = 11 \text{ L-m} \text{ S} \text{ S} \text{ S}^{\circ} \text{ C} = 26^{\circ} \text{ F}$ - Rw = 22 1-4 0 50°E (b) by Chencel andyris 2cat=10.69 2an = 11.30Rom Ainstillelson p Z6 R (du-u) = 100 Z ce nog R/25= 10-69+11.3 = 4.6 - N-m/2500 icho $Rw \Big|_{T2} = Ra \Big|_{T_1} = \frac{T_1 + 6e T7}{T_2 + 6e T7}$ $\frac{R_{50}}{50} = \frac{77 + 8.77}{50 + 6077} = 4.6 \times 1.48 = 6.8 \text{ A-m}$ Rw= 6.8 1-4 1507 or pathops Ce = 11 $Are P_{2s} = \frac{100}{11} = 9.1 - 4 - 41$ $\frac{R}{R} = 2.1 + 1.48 = 13.5$

ار بالاست. از مسید مشکل میکند است است که به این م

 $\phi = \sqrt{\frac{R_w}{R_0}} = \sqrt{\frac{14}{70}} = \frac{0.45}{2.45} = \frac{45\%}{45}$ Duis is much tes high (b) $\frac{1}{2} - \frac{1}{2} - \frac{1}{2} = 0.56$ (C) For Rw= 6.8 as per chen andyses $\varphi = \begin{pmatrix} \frac{6 \cdot 8}{70} = 0.31 \\ - \frac{1}{70} \end{pmatrix}$ $\phi = \sqrt{\overline{3}} = 0.37$

LITTLE CRATER CAMPGROUND WARM WELL

• .

SPECIES	CONCENTRATION (ppm)	ANALYTICAL METHOD	DETECTION LIMITS	COh	CENTRATION
Na	83.00	1.	• 61		.361E-02
K	10.00	1	1.22		.256E-03
Ca	54.00	1	- 24		·135E-02
Mg	48.00	1	.49		∎197E-02
Fe	4.00	1	• O2		•716E-04
A1	N.D.	1	• 6 1	<	.741E-07
Si02 -	161.00	1	• 52		.268E-02
B	2.50	1	-12		.231E-03
Li	-12	1.	.05		.173E−Ođ
Sr	N • D •	1	• O 1	<	" OOOE +OO
Zn	N . D .	1	-12	<	• 000E+00
Ag	N.D.	1	.05	<	•000E+00
As	N . D .	1	= 6: 1	<	•000E+00
Au	N . D .	1	1 O	\sim $<$	• 000E + 00
Ba	N.D.	1	• 61	<	.000E+00
Be	N - D -	1	•00	<	*0005+00
Bi .	N.D.	1	2.44	<	.000E+00
Cd	N.D.	1	.06	<	.000E+00
Ce	N.D.	1	• 2d	<	.000E+00
Co	N.D.	. 1	•02	<	•000E+00
Cr ·	N.D.	1	05	<	.000E+00
Cu	N • D •	1	• 06	<	•000E+00
La	N.D.	1	• 1 2	<	• 000E+00
Mn		1	• 24		.4558-05
Mo	N • D •	1	1.22	<	∎000E+00
Ni	N • D •	1	-12	<	•000E+00
РЬ	N - D -	1	• 24	<	"000E+00
Sn	N • D •	1	-12	<	•000E+00
Sb	N - D -	. 1	.73	<	• 000E+00.
Te	N.D.	1	1.22	<	.000E+00
Th	N.D.	1	2.44	<	•000E+00
Τi	N • D •	1	.12	<	•000E+00
L)	N.D.	1	6.10	<	• 000E+00
V	N.D.	1	1.22	<	• 000E +00
W	N - D -	1	-12	<	*000E+00
Z r	N.D.	1.	· · 12 .	<	•000E+00

1



1

180 CATION RETENTION

is minimized if the index solution is lowered to approximately 0.1 M during the final two saturation washes. The error is eliminated if the quantities of index salt are analytically determined instead.

ANNNOTATED BIBLIOGRAPHY

- Babcock, K. L. 1963. Theory of the Chemical Properties of Soil Colloidal Systems at Equilibrium. *Hilgardia* 34:417-542. Excellent section on various cation-exchange equations, their usages and implications.
- Bolt, G. H. 1955a. Analysis of the Validity of the Gouy-Chapman Theory of the Electric Double Layer. J. Colloid. Sci. 10:206-218. Good discussion of assumptions and potential errors in standard theory of the diffuse double layer.
- Bolt, G. H. 1955b. Ion Adsorption by Clays. Soil Sci. 79:267-276. Popularization and further testing of Eriksson's diffuse double layer exchange equation.
- Bower, C. A. 1959. Cation-Exchange Equilibria in Soils Affected by Sodium Salts. Soil Sci. 88:25-35. Detailed discussion of problems encountered in cation exchange measurements, and their solutions.
- Eriksson, E. 1952. Cation Exchange Equilibria on Clay Minerals. Soil Sci. 74:103-113. Development of diffuse double layer exchange equation in its common form.
- Gast, R. G. 1977. Surface and Colloid Chemistry. In Minerals in Soil Environments.
 J. B. Dixon and S. B. Weed (Eds.). Soil Science Society of America, Madison, Wisconsin.
- Haan, F. A. M. de. 1965. The Interaction of Certain Inorganic Anions with Clays and Soils. Centrum voor Landbouwpublikaties en Landbouwdocumentatie, Wageningen. 167 pp. Excellent mathematical and experimental treatment of anion repulsion and adsorption by soils.
- Kelley, W. P. 1948. *Cation Exchange in Soils*. American Chemical Society Monograph No. 109. Reinhold, New York. 144 pp. Good early summary of cation exchange in soils.
- Kemper, W. D. and W. P. Quirk. 1970. Graphic Presentation of a Mathematical Solution for Interacting Double Layers. Soil Sci. Soc. Amer. Proc. 34:347-350. Realistic attempt to simplify diffuse double layer computations, including sample calculations.
- Nielsen, D. R., R. D. Jackson, J. W. Cary, and D. D. Evans, (Eds.). 1972. Soil Water. American Society of Agronomy, Madison, Wisconsin. 175 pp. An abbreviated treatment of diffuse double layer theory, with excellent summary of status of the theory at the time of publication and a perspective on its utility from the standpoint of soil physics.
- Okazaki, R., H. W. Smith, and C. D. Moodie. 1963. Hydrolysis and Salt Retention Errors in Conventional Cation-Exchange-Capacity Procedures. Soil Sci. 96:205-209. Detailed discussion of problems inherent in standard cation exchange measurements.

Overbeek, J. T. G. 1952. Electrochemistry of the Double Layer. In Colloid Sci. 1:115-193. H. R. Kruyt, (Ed.). Detailed treatment of diffuse double layer theory.

Shainberg, I. and W. D. Kemper. 1966. Hydration Status of Adsorbed Cations. Soil Sci. Soc. Amer. Proc. 30:707-713. Excellent sample of calculations involved in dealing with ion-exchange selectivity.

Thompson, H. S. 1850. On the Absorbent Power of Soils. J. Royal Agr. Soc. 11:68-74. One of the earlier reports on cation exchange phenomena in soils.-

- Van Olphen, H. 1963. An Introduction to Clay Colloid Chemistry. Interscience, New York. 301 pp. Simplified treatment of diffuse double layer theory, stressing interparticle bonding. Includes sample calculations.
- Way, J. T. 1850. On the Power of Soils to Absorb Manure. J. Royal Agr. Soc. 11:313-379. Another early report on cation exchange phenomena in soils.

QUESTIONS AND PROBLEMS

1. The following distribution of cations and anions exists near a soil colloid surface:

	4.0 nm	3.0 nm	2.0 nm	1.0 nm	0.5 nm	0.25 nm
Cation concentration						
$(mole (+) L^{-1})$	0.10	0.12	0.17	0.35	1.0	2.0
Anion concentration						
$(mole(-)L^{-1})$	0.10	0.08	0.06	0.04	0.01	0.00

Assuming that the excess of cations reported for each increment represents the entire increment (e.g., that the cation concentration is 2.0 moles charge L^{-1} from the colloid surface to 0.375 nm from the surface, etc.), estimate the CEC for a colloid having $800 \times 10^3 \text{ m}^2 \text{ kg}^{-1}$ of reactive surface (Ans. = 12.0 mmoles charge kg⁻¹).

- 2. Based upon the data of Table 6.4, what proportion of the cross-sectional area of a cylindrical soil pore of radius 15 μ m is influenced by the electric double layer if monovalent ions predominate at a salt concentration of 10^{-1} M?
- 3. If all water of a desaturated soil at 20% water content is spread uniformly over $100 \times 10^3 \text{ m}^2 \text{ kg}^{-1}$ of reactive surface, what proportion of that water is influenced by the electric double layer for the chemical conditions specified in Problem 2?
- 4. A soil is equilibrated with a solution of SAR = 20. Based upon the Gapon equation, what would be its equilibrium exchangeable sodium per-

proximately 0.1 *M* during the ated if the quantities of index

erties of Soil Colloidal Systems at ection on various cation-exchange

iouy-Chapman Theory of the Elecood discussion of assumptions and e double layer.

ci. 79:267-276. Popularization and r exchange equation.

in Soils Affected by Sodium Salts. ems encountered in cation exchange

on Clay Minerals. Soil Sci. 74:103ange equation in its common form.

1. In Minerals in Soil Environments. ence Society of America, Madison,

tain Inorganic Anions with Clays and en Landbouwdocumentatie, Wal experimental treatment of anion re-

. American Chemical Society Monop. Good early summary of cation ex-

phic Presentation of a Mathematical it Sci. Soc. Amer. Proc. 34:347-350. layer computations, including sample

d D. D. Evans, (Eds.). 1972. Soil Wan, Wisconsin. 175 pp. An abbreviated ith excellent summary of status of the sective on its utility from the standpoint

ie. 1963. Hydrolysis and Salt Retention Capacity Procedures. Soil Sci. 96:205therent in standard cation exchange