ISOTOPIC COMPOSITIONS OF

, Department of Geosciences, The arsity Park, Pennsylvania 16802 ng fluid-rock interactions affected by equilibrium temperaa the hydrothermal fluid.  $\delta D$ y: (1) source of the hydro-; (2) isotopic exchange history the physico-chemical state of e dependent on (3) because any lter both the relative propor-CH4, CO2, CO), and the isotopic

e isotopic fractionation factors inges in T, Pfluid, fo2, and isotopic equilibria among fluid eologically important  $f_{02}$  region aral buffers W-M, and M-H), a se large changes in δD and/or d from the fluid. For example, 18 O values of minerals formed ffer by more than 30% and 5%, tate of the fluid. This implies igin of fluids from  $\delta D$  and  $\delta^{18} O$ physico-chemical parameters tion can be estimated.

## THOPHORE DISTRIBUTION

drew, Lamont-Doherty Geoloniversity, Palisades, New

communities were studied at approximately 600 samples 0m and 100m. Significant anstations in terms of the stander temperature. The seasonne rally similar to each other ed, although, occasional barren rgest standing crops were as-. Most species occured only omenon was repeated year each species, however, varied ption of Emiliania huxleyi . As already shown in the tions and the seasonal varia-100m level from that of the eved, a fairly large number of d water, and more detailed imum and optimum water tem-

In addition to the semi-

monthly samples, some semiweekly and daily samples were also collected to examine the short term variations of the flora. While species composition was found to be relatively consistant within these samples, the standing crop showed great variation.

VADOSE PRECIPITATION OF HIGH MAGNESIAN CALCITE AND ARAGONITE IN THE MONARCH MINE, KICKING HORSE PASS, BRITISH COLUMBIA

Oldershaw, A. E.; McIlreath, I. A.; Ghent, E. D.; Department of Geology, University of Calgary, Alberta, Canada Dolomite clasts on the floor of the East Monarch Mine are undergoing rapid lithification by precipitation of high-magnesian calcite and aragonite. Precipitation is taking place at present and appears to have been continuous for a maximum of 70 years and a minimum of 20

High-magnesian calcite, aragonite and intergrowths of these two minerals form alternating, roughly concentric bands around angular, grain-supported dolomite clasts. The resulting fabric is comparable to that of vadose pisoliths described from other localities, including the development of pendulous rims on the lower surfaces of the clasts and the occurrence of inverse grading of the "pisoliths".

The high-magnesian calcite (average 12 mol% MgCO3 by microprobe) exhibits a blocky sub-rhombic crystal form with maximum crystal diameters ranging from 0.5 µ to 3.0 µ. Aragonite, identified by XRD, contains an average of 1  $\mathrm{mol\%}\ \mathrm{MgCO_3}$  and occurs as twinned, fibrous to bladed crystals from 0.25µ to 3.0µ wide and 10µ to 50µ long: the aragonite exhibits both an oriented and a tangential fabric relative to the clast surfaces. In the mixed zones, aragonite adopts a rosette form and the high-magnesian calcite tends to be more equicrystalline and of a finer crystal size.

The source of the precipitating carbonates is inferred to be meteoric ground waters percolating through an overlying dolostone source rock. Variations in mineralogy and crystal morphology can be explained by semi-seasonal fluctuation in ground water chemistry, particularly magnesium ion concentration.

LEACH HOT SPRINGS GEOTHERMAL AREA, NEVADA

Olmstead, F. H., U. S. Geological Survey, Water Resources Division, Menlo Park, California; Van Denburgh, A. S., U. S. Geological Survey, Carson City, Nevada

Leach Hot Springs are in southeastern Grass Valley, about 45 km by road south of Winnemucca, Nevada. The springs issue from steeply inclined fault-controlled conduits in Quaternary alluvium and Tertiary nonmarine sedimentary rocks. Late Paleozoic and early Mesozoic sedimentary and volcanic rocks are exposed east of the springs and probably underlie the spring area at depths of no more than a few hundred metres. Total discharge of thermal water is about 3-5 x 105 m<sup>3</sup> yr<sup>-1</sup>, most of which is springflow from 30 orifices. Discharge temperatures in November 1973 ranged from 34.5°C to 95.5°C (boiling); the weighted average was 76°C. Dissolved solids, dominated by sodium, bicarbonate, and silica, total  $_{
m about}$  580 mg  $1^{-1}$ . Estimated reservoir (equilibrium) temperature is 155°C by the silica-quartz geothermometer or 176°C by the sodium-potassiumcalcium geothermometer. The thermal anomaly associated with the spring system is defined by temperatures greater than about 15°C at a depth of

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30 m below land surface. The anomaly forms an ellipse with the major  $_{\rm axis}$  oriented northeastward, along the fault that controls the position of the hot springs. Comparison of the temperature pattern with the configuration of the water table indicates that transport of heat by lateral flow of thermal ground water is not significant at shallow depths. Total discharge of heat from the system is about 1.8 x  $10^6$  cal s $^{-1}$ , of which 1.0 x  $10^6$  cal s $^{-1}$  is conduction through near-surface materials within the area of the thermal anomaly and the remainder is the heat carried by the spring discharge.

GEOTHERMAL RESOURCE CHARACTERISTICS,

EXPLORATION CONSIDERATIONS AND PROPERTY ACQUISITION TECHNIQUES

Harry J. Olson and William M. Dolan, Amax Exploration, Denver, Co. World wide experience demonstrates rather conclusively that geothermal reservoirs are more closely related as far as geological occurrence, and exploration and development techniques are concerned to conditions encountered by the mining industry than thoseencountered by the petroleum industry. To date, however, the majority of geothermal activity in the United States has been undertaken by the petroleum industry or related organizations not only because of the energy relationship but also because property acquisition practices and overall exploration costs have a more direct bearing to the oil and gas experience

Due to high land acquisition and evaluation costs, and because of long developmental lead time, complex marketing problems and the necessity to develop a technical capability to handle a wide range of geothermal fluids, the high degree of inherent risk in geothermal ventures requires that costs related to the discovery and definition of the resource be held to a minimum. This can be accomplished by avoiding large lease bonuses and rentals for unevaluated and untested properties, by joint venture arrangements whereby a degree of the risk is spread, and by exploration agreements that permit a timely evaluation of the geothermal potential without a committment to large capital expenditures.

CHEMICAL INTERACTIONS DUE TO SUBSURFACE MIXING OF METEORIC AND MARINE WATERS IN A PLEISTOCENE REEF COMPLEX, RIO BUENO, JAMAICA

O'Neil, Thomas J., Department of Geological Sciences, University of Southern California, Los Angeles, California, 90007

Southern California, Los Angeles, California, 90007
An investigation was made of geochemical processes occurring during subsurface mixing of meteoric and marine waters within a Pleistocene reef complex along the north coast of Jamaica. Water samples from local sea water, four shallow wells along a transect normal to the strand line, and a fresh water spring were monitored over a four-week period for Na, Mg, Ca, K, Sr, Cl, SO<sub>4</sub>, pH, and alkalinity.

Na, K, Cl, and SO<sub>4</sub> behave conservatively during mixing and decrease in concentration inland toward fresh water recharge zones. Variations in Ca, Mg, Sr, pH, and alkalinity, however, reflect variations occurring between water and sediment in the zone of mixing. Thermodynamic calculations indicate that meteoric water is saturated with respect to magnesian calcite and slightly undersaturated to aragnite and dolomite. Waters in the zone of mixing are slightly undersaturated with respect to magnesian calcite and aragonite, but supersaturated to dolomite. Solution cavities in sediments of the mixed zone reflect dissolution by waters undersaturated in magnesian calcite

and aragonite. Further samp of dolomite cements.

Decreased saturation of carbonate solid phases result coefficients and degree of cochanges in ionic strength.

THE MAGNETIC POLARITY STRATIG SEQUENCE AT ANZA-BORREGO STAT Opdyke, N.D., Lamont-Doher York 10964; Lindsay, E.I

85721; Johnson, N.M., Dai 03755; Downs, T., Museum Angeles, California 9000 One hundred and twenty sites withrough a stratigraphically commammal bearing poorly indurate The magnetically oriented sampthe clays and fine silts. A.f

500 oersteds peak field for its Eleven complete reversals section which contains fossils tonian. On the basis of the copreviously studied section in the Arizona, the observed magnetic represent the time from the mid the upper Matuyama reversed policepresented by 4500 feet of secfor this part of the sequence of

samples and it was found that

is present in most samples req

PALEOHYDROLOGY OF A CARBONIFEROU MOROCCO

Padgett, Guy V., Department of Columbia, S.C., 29208 oint bar accretion scars expose of southern Morocco offer an exc eleohydraulics of these deposits baleo-drainage basin. Gently di . ipped of overlying siltstone a urface for several kilometers. ially eroded, removing the highl ar, resulting in a series of sha rined a point bar ridge. Radii idges were measured from aerial aleodischarge, drainage area, me ne meander wavelength thus gener avelength observed and measured verall thickness of the point bar f mean velocity was calculated. ere repeated at several locations on. A geographic plot of the res osaic for this late paleozoic sed