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CHEMISTRY, ORIGIN, AND POTENTIAL OF GEOTHERMAL RESOURCES IN SOUTHWEST NEW MEXICO AND SOUTHEAST ARIZONA

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Chandler A. Swanberg

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ABSTRACT

On the basis of chemical geothermometry, there would appear to be 10-20 geothermal prospect areas in southwest New Mexico and southeast Arizona whose reservoir base temperature approaches the 150°C minimum for economic generation of electricity. The most promising of these prospect areas are either associated with Quaternary volcanics or are located along the margins of the deepest sedimentary basins. The thermal waters tend to be slightly saline (1000-3000 mg/Å) and enriched in silica and fluoride. The numerous hot spring areas in the Gila National Forest of southwest New Mexico do not appear to have a sufficiently high reservoir base temperature for electricity generation. However, the low salinity (<500 mg/Å) and high surface discharge temperature (up to 75°C) of these hot springs make them ideal for non-electric applications.

INTRODUCTION

To date, we have visited nearly all hot springs in Arizona and New Mexico, recorded the temperature and collect samples for chemical analysis. In addition, we have examined the chemistry of several thousand non thermal ground waters to establish background chemistry for comparison against thermal water chemistry. Standard methods of quantitative and qualitative geothermometry (see Truesdell, 1975 for a summary of techniques) have been applied to all waters and the resulting geotemperatures used to predict the subsurface temperature anticipated for each geothermal prospect area. The most promising geothermal areas are designated in Figure 1. Table 1 contains the chemistry of selected thermal waters from southwest New Mexico and southeast Arizona.

Geothermometry Techniques

The concept of chemical geothermometry is that the chemistry of geothermal fluids is controlled by temperature dependent waterrock reactions within the geothermal reservoir and that the water chemistry does not change appreciably as the water migrates from the geothermal reservoir to the surface sampling point. The validity of these assumptions is examined by Truesdell (1975).

The silica geothermometer of Fournier and Rowe (1966) can be quantatively expressed according to the equation of Truesdell (1975)

$$T SiO_2 = \frac{1315}{5.205 - \log_{10} SiO_2} - 273.15$$
(1)

where T SiO_2 is the silica geotemperature in degrees Celcius, and SiO_2 is expressed in parts per million. The NaKCa geothermometer of Fournier and Truesdell (1973) can be quantatively expressed according to the equation of Truesdell (1975)

$$T_{NaKCa} = \frac{1647}{log_{10}(Na/K) + \beta log_{10}(\sqrt{Ca}/Na) + 2.24} - 273.15$$
(2)

where T NaKCa is the NaKCa geotemperature in degrees Celcius, Na, K, and Ca are expressed in molal concentrations, and the value of β is determined by the following tests:

$$\beta = 4/3$$
 for $\sqrt{Ca/Na} > 1$ and T NaKCa($\beta = 4/3$) < 100°C
 $\beta = 1/3$ for $\sqrt{Ca/Na} < 1$ or T NaKCa($\beta = 4/3$) > 100°C.

The normal distribution of geotemperatures calculated by applying equations 1 and 2 to groundwaters of the Basin and Range province is given in Figures 2,3. The geotemperatures obtained by applying equations 1 and 2 to selected thermal waters from Arizona and New Mexico are given in Table 1.

Discussion

The most promising geothermal prospect areas on the basis of chemical geothermometry are shown in Figure 1. Many of these prospects are located along the flanks of the deep sedimentary basins of the Rio Grande Rift, an association which implies a tectonic origin for these thermal waters. These waters have apparently originated deep within the basin where they have been heated by the normal geothermal gradient, and then migrated to the surface along the basin bounding faults. If this is the case, the waters may have traveled a considerable lateral distance enroute to the surface so that the locations given in Table 1 and shown in Figure 1 may not accurately represent the location of the subsurface geothermal area.

There are at least ten separate localities in the Pio Grande Rift of southern New Mexico where geothermal waters leak to the surface, most of which are not designated individually in Figure 1.

An additional four areas in west Texas are also shown to emphasize the association with deep sedimentary basins and active faults but are not treated further. These waters are characterized by NaKCa geotemperatures in the 150-200°C range and silica geotemperatures in the 80-120°C. The silica data do not appear to reveal much geothermal potential until it is realized that these waters have low silica because they have ascended through a very thick pile (see Figure 1) of sediments saturated by the silica deficient waters of the Rio Grande River. Application of mixing models to the Radium Springs data brings the silica geotemperature into agreement with the NaKCa geotemperatures. Thus most of the Rio Grande Rift thermal areas are likely to have a reservoir base temperature near or in excess of the 150°C minimum for economic generation of electricity.

A second group of thermal waters worth special mention are those located in the Gila area just west of the Rio Grande Rift in southwest New Mexico. Chemical analyses for several of the hottest springs are present in Table 1 but are omitted from Figure 1 as their NaKCa and silica geotemperatures are not sufficiently above regional background (Figures 2,3; see also Swanberg and Alexander, in press) to suggest the presence of a buried geothermal resource. However, these springs have a high surface discharge temperature (up to 75°C), very low amounts of dissolved solids (<500 mg/L) and are quite numerous so that the Gila area is ideal for non electric applications of geothermal energy.

Two exceptions to the above generalizations are the Lower Frisco hot springs and the Clifton Known Geothermal Resource Areas (KGRA), located on either side of the New Mexico-Arizona border at about

parallel 33°N. Both areas appear to have subsurface temperatures of about 150°C (Table 1).

A final feature of geothermal resources in southwest New Mexico and southeast Arizona is the presence of very promising geothermal prospects associated with Quaternary volcanic centers (Figure 1). Examples include Kilbourne Hole, New Mexico, east of Douglas, Arizona, and although it is not included in Figure 1, the Springerville area of Arizona. All of these areas are characterized by sodium bicarbonate water, and Na K Ca and silica geotemperatures near 150°C.

REFERENCES

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- Swanberg, C.A. and Alexander, S., in press, The use of the water quality file, WATSTORE in geothermal exploration: An example from the Imperial Valley, California: Submitted to Geology.
- Truesdell, A.H., 1975, Summary of section III: Geochemical techniques in Exploration: Proc. 2nd U.N. Symp. on the Development and Use of Geothermal Resources, San Francisco, Calif., USA, 20-29, May, 1975, U.S. Govt. Printing Office, p. LIII-LXXIX.

FIGURE CAPTIONS

- Figure 1. Association among high temperature geothermal resources (>150°C), active faults, deep sedimentary basins, and Quaternary basaltic volcances. All data except the geothermal areas are from a forthcoming paper by W. Seager and P. Morgan on the Rio Grande Rift (J.G.R. in press) and is reproduced with the kind permission of the authors.
- Figure 2. Histogram of T NaKCa obtained by applying equation 2 to groundwaters of the Basin and Range province.
- Figure 3. Histogram of T SiO₂ obtained by applying equation 1 to groundwaters of the Basin and Range province and the Imperial Valley. After Swanberg and Alexander (in press).





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		Location																						
Region	Sample		Latitude		Longitude		Тетр.	Temp. Na-Ca-K	Temp. SiO ₂	TDS	рH	Na	ĸ	Ca	Fе	Мg	C0 ₃	нсө _з	C1	50 ₄	F	В	S10 ₂	Conment
Río Grá	ande Rift	,		· -																				
	256	Columbus Area	310	47.7		34.3	26.0	191.80	136.4 ⁰	2748	7.36	881.0	75.4	65.2	.82	19.8	0	1565.7	338.9	423.6	3.99	1.78	98.5	
	B2	Radium Springs	32	30.1	106	55.7	53.0	223.4	118.1	3532	8.16	1135.9	167.0	118.6	. 49	15.2	13,2	378.3	1693.6	263.2	4.44	.86	69.9	KGRA
	85	Derry Springs	32	47.7	107	16.6		156.2	81.2	1240	8.23	323.9	38.8	47.1	<.10	15.8	Û	366.1	151.0	376.6	5.90	. 34	31,3	
	610	Truth or Consequences	33	8.1	107	14.9	41.0	178.1	96.2	2688	7.80	791.5	53.0	143.9	<.10	17.9	C	162.3	1353.6	169.1	3.10	.35	44.3	
	Bl3	Sedillo Springs	34	2.2	105	56.2	34.0	58,5	72.5	284	8.48	56.1	3.1	17.2	<.10	4.3	0	162.3	10.3	50.0	2.02	.09	25.3	KGRA
		San Diego Hountain	32	37.0	106	58.0		232.5	105	4970	7.9	150.0	210	55	.04	5.3	0	675	880	1900	17.0	1.0	53.0	
		S. Tularosa	32	8.0	106	8.0	71.1		109	8740				542		108			6590	820			58.0	
		Charberino	32	5.0	106	37.0	20	198	101	1400	7.5	300	45	110	.70	43	۵	456	410	210	.3		49	
Gila A	rea	<u> </u>																						
	29	Faywood Spring	32	33.3	107	59.7	58.8	78.4	97.2	492	7.74	90.8	8,2	35.6	.12	7.6	0	283.0	14.2	72.0	6.10	101	45.2	
	31	Mimores Spring	32	44.9	107	50.1	58.2	74.5	106.8	320	8.97	91.7	1.2	2.4	<.10	0	20.4	67.1	14.5	84.0	16.0	3	55.6	
	33	Gila Spring	33	12.0	108	12.6	66.3	77.3	120.5	416	8,15	129.7	3.1	10,4	<.10	.2	0	115.9	100.1	67.2	8.60	.02	73.3	KGRA
	150	Turkey Creek	33	6.5	108	24.0	74.6	56.2	115.5	236	8.66	61.1	1.5	6.8	.31	1.6	0	94.0	4.2	64.8	9.45	.12	67.7	
Quater	namy Basi	elts																						
	W33	Kilbourne Hole	e 31	47.3	107	1.7		174.7	133.5	720	9.42	233.3	16.8	10.0	<.15	2.1	69.6	371.0	29.4	101.8	3,30	. 49	93,5	KGRA
	338	Douglas	31	21.1	109	10.9	25.5	229	129	784	9.06	200.0	37,9	9.8	4.53	45.9	70.8	588.2	20.9	64.4	2.48	, .11	86.5	
	112	Springerville	34	22.1	109	23.0	18	192		300	8,92	79.3	10.2	6.0	.30	5.1	0	107.4	81.2	5.8	2,48	; .10		
Basin	and Rang	ē																						
	P2	Lightning Doc	k 32	8.7	108	49.9	85,0	172.9	160.1	1116	7,71	333.6	23.5	22.0	. 20	0.5	0	106.8	88.3	497.1	12.6	; .48	147.5	KGRA
	J5	Lower Frisco	33	14.5	108	52.8	48.9	147.9	131.9	1280	7,79	406.0	18.8	54.3	· <.10	5.9	0	107.4	574.3	90.3	1.8	30.38	90.4	KGRA
	ŝ	Clifton H.S.	33	4.8	109	17.8	48.0	180	153	14548	7.86	3585.9	243.9	925.8	.73	2 22.9	ũ	150.0	7484,5	0	3.	501.51	131.4	Кыка
	7	Gillard H.S.	32	58.4	104	20.9	82.0	139	136	1244	8.04	410.8	13,2	20.0	<.1	a.7	۵	219.6	463.6	174.7	10.	50,40	37.9	KOKA
	11	Indian H.S.	33	0.2	109	54.0	46,5	100	95	3004	7.88	1022.6	12.9	92.8	3 <.1	0 10.3	0	101.2	202 0	361.0) 3.5 2 10	80.70 6.13	43.5 66 9	
	15	Safford	32	50.7	109	33.6	43.5	106	116	1076	8.54	330.7	4.	3 7.4	· .]	0 1.3 	3.2 0	233.0	203.0	293.0	יטי נ. ארו נ	0J 00 85	39 K	
	133	Larasburg	32	13.7	108	30.7	33.0	150.6	91.2	816	7.86	5 216.1	11.	/ 28.0	I	5 2.7	U	314,8	47.3	233.6	. U.	10 .40	39.0	

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TABLE 1: Chemistry of Selective Geothermal Waters from Southwest New Mexico and Southeast Arizona.