## INTER OFFICE MEMORANDUM

Date: February 18, 1975

To: W. M. Dolan R. F. Horsnail H. J. Olson A. L. Lange

H. D. Pilkington 🖌

M. H. Alldredge

From: Frank Dellechaie

Subject: Preliminary report on the thermal feature of the Beulah Reservoir area, Oregon

Thermal Features

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Eighteen water samples were collected in the vicinity of Beulah reservoir, Oregon, during July and August, 1974. Water temperatures vary from 12°C at Main Cold Spring to 72°C at Beulah Hot Spring. The background temperature is about 14°C. The Beulah reservoir spring issues out of Miocene and Pliocene vitric tuffs on the north east shore of Beulah reservoir. The springs have deposited . large quantities of siliceous sinter and continue to do so. Hot springs also issue from the north fork of the Malheur River 3 miles WNW of the Beulah Hot Springs and one mile north of Juntura. Spring or well names, temperature and flow rates are listed in Table 1.

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Table 1. Thermal features of the Beulah Reservoir area.

Name	T°C	Discharge gpm
Beulah Hot Spring #1	72	10
Beulah Hot Spring #2	72	50
Section 29 Warm Spring	44.5	15
Bath House Hot Spring	35	6
Butler Warm Well	25	50
De Armond Warm Spring	20.5	10

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## Chemistry

Thermal waters in the Beulah reservoir area are of the sodiumsulfate-bicarbonate variety (see Table 2). Chloride concentrations are inordinately low for a hot water system and approachesthe 10 mg/l chloride boundary that separates hot water and dry steam systems. White, et al (1971) report springs associated with vapor-dominated systems that have near neutral pH, high sodium and bicarbonate and low chloride. Cl/F ratios range from 2.2 to 7.4, unusually low for a hot water system. Cl/SO<sub>4</sub> ratios are very low and consistently show more SO<sub>4</sub> than Cl. B, Li, NH<sub>3</sub> and H<sub>2</sub>S are all volatile and are found in favorably high levels in the Beulah Hot Springs water.

Silica thermometry corrected for amorphous silica yields temperatures of about 125°C. Na/K temperatures show unreasonably low temperatures i.e. less than 68°C. Na-K-Ca temperatures correlate well with silica temperatures. The water from Beulah Hot Spring appears to have equilibrated at about 125°C.

Two interpretations are offered to explain the chemistry of Beulah Hot Springs. First, the chemistry may be interpreted as representing a rather luke-warm hot water system that would not offer temperature in excess of 125°C. Secondly, the chemistry may represent a hot water system on the periphery of a dry steam system. The second interpretation is based on the favorable concentrations of volatile elements and the conspicuous dearth of chloride. An analysis of a sample from the deepest wells of the Carboli hot water field located on the southern border of the Larderello steam field by Cataldi et al (1969) is included in Table 2 for comparison.

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Table 2. A chemical comparison of a typical cold spring and hot springs from the Beulah Reservoir Area, Oregon, and a hot well from Carboli, Italy. Concentration units are mg/l unless otherwise noted.

	Castle Rock Cold Spring	Malheur River Hot Spring	Beulah Hot Spring_#1	Section 29 Warm Spring	Italy Cataldi et al 1969
рH	8.10	8.50	7.60	9.6	≈ 7
Cl	3.6	20	51	11	42.6
Z	0.2	4.8	4.8	0.8	<b>—</b> —
HCO <sub>3</sub>	90	100	160	50	89.7
CO3	0	40	0	48	<del></del>
SOu	4	70	290	38	137.4
Si0 <sub>2</sub>	70	82	194	70	
Na	1.2	120	220	72	56.6
K	3.8	1.6	5.8	0.7	32
Ca	16	3	18	1	
Mg	7	0.1	0.3	<0.1	5
Li	<0.1	0.21	0.24	<0.1	
В	<1.0	1.1	4.6	<1.0	13.9
Cu	<0.1	0.1	0.1	<0.1	
Mo (ug/l)	<1	15	30	8	
Zn	0.1	0.1	0.1	0.2	
Fe	<0.1	<0.1	<0.1	<0.1	
NH <sub>3</sub>		0.49	1.3	·	19
H2Š		<b></b>	0.78		·
TDS	208	444	952	293	
T°C	16.5	58	72	44.5	300
Flow gpm	50	50	10	15	
TSiO <sub>2</sub> (Total) °C	118.2	126.4	177.9	118.2	
TSiO <sub>2</sub> (Amorphous) °C	2.9	· 4.4	52.4	-2.9	
TSiO <sub>2</sub> (Quartz) °C	118.2	122.0	125.5	118.2	
TNa/Ř °C	374.2*	28.5	67.7	13.2	
TNa-K-Ca °C	51.6	100.1	122.8	89.3	
Cl/SO <sub>L</sub>	2.4	0.77	0.48	0.78	
Cl/F <sup>4</sup>	9.6	2.2	5.7	7.4	
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\* Does not reflect subsurface equilibrium.